AMMONIUM BROMIDE

1. Product Identification

Synonyms: Hydrobromic acid monoammoniate  
CAS No.: 12124-97-9  
Molecular Weight: 97.95  
Chemical Formula: NH4Br  
Product Codes:  
J.T. Baker: 0636  
Mallinckrodt: 0424

2. Composition/Information on Ingredients

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>CAS No</th>
<th>Percent</th>
<th>Hazardous</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ammonium Bromide /(NH4)Br/</td>
<td>12124-97-9</td>
<td>100%</td>
<td>Yes</td>
</tr>
</tbody>
</table>

3. Hazards Identification

Emergency Overview

WARNING! HARMFUL IF SWALLOWED OR INHALED. AFFECTS CENTRAL NERVOUS SYSTEM, BRAIN AND EYES. MAY CAUSE IRRITATION TO SKIN, EYES, AND RESPIRATORY TRACT.

SAF-T-DATA(tm) Ratings (Provided here for your convenience)

Health Rating: 2 - Moderate (Life)  
Flammability Rating: 0 - None  
Reactivity Rating: 2 - Moderate  
Contact Rating: 2 - Moderate  
Lab Protective Equip: GOGGLES & SHIELD; LAB COAT & APRON; VENT HOOD; PROPER GLOVES  
Storage Color Code: Green (General Storage)
Potential Health Effects
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Inhalation:
Dust may cause irritation to the respiratory tract. Symptoms may include coughing, sore throat, and shortness of breath.

Ingestion:
May cause nausea, vomiting and abdominal pain. Ingestions are usually promptly rejected by vomiting, but sufficient absorption may occur to produce central nervous system, eye and brain effects. Symptoms may include skin rash, blurred vision and other eye effects, drowsiness, irritability, dizziness, mania, hallucinations, and coma.

Skin Contact:
Dry material may cause mild irritation. Solutions may cause irritation, redness, pain, and skin burns.

Eye Contact:
May cause irritation, redness and pain.

Chronic Exposure:
Repeated or prolonged exposure by any route may cause skin rashes (bromaderma). Repeated ingestion of small amounts may cause central nervous system depression, including depression, ataxia, psychoses, memory loss, irritability, and headache.

Aggravation of Pre-existing Conditions:
Persons suffering from debilitation, depression, alcoholism, neurological or psychological disorders may be more susceptible to the effects of this compound.

4. First Aid Measures

Inhalation:
Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.

Ingestion:
Induce vomiting immediately as directed by medical personnel. Never give anything by mouth to an unconscious person. Call a physician.

Skin Contact:
In case of contact, immediately flush skin with plenty of water for at least 15 minutes. Call a physician.

Eye Contact:
Immediately flush eyes with plenty of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Get medical attention immediately.

5. Fire Fighting Measures

Fire:
Not considered to be a fire hazard.

Explosion:
Not considered an explosion hazard, but reacts explosively with bromine trifluoride.

Fire Extinguishing Media:
Use any means suitable for extinguishing surrounding fire. Water spray may be used to keep fire exposed containers cool.

Special Information:
In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode.
6. Accidental Release Measures

Ventilatate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Spills: Sweep up and containerize for reclamation or disposal. Vacuuming or wet sweeping may be used to avoid dust dispersal.

7. Handling and Storage

Keep in a tightly closed container, stored in a cool, dry, ventilated area. Protect against physical damage. Isolate from acids and alkalis. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.

8. Exposure Controls/Personal Protection

Airborne Exposure Limits:
None established.

Ventilation System:
A system of local and/or general exhaust is recommended to keep employee exposures as low as possible. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, *Industrial Ventilation, A Manual of Recommended Practices*, most recent edition, for details.

Personal Respirators (NIOSH Approved):
For conditions of use where exposure to dust or mist is apparent and engineering controls are not feasible, a particulate respirator (NIOSH type N95 or better filters) may be worn. If oil particles (e.g. lubricants, cutting fluids, glycerine, etc.) are present, use a NIOSH type R or P filter. For emergencies or instances where the exposure levels are not known, use a full-face positive-pressure, air-supplied respirator. WARNING: Air-purifying respirators do not protect workers in oxygen-deficient atmospheres.

Skin Protection:
Wear protective gloves and clean body-covering clothing.

Eye Protection:
Use chemical safety goggles. Maintain eye wash fountain and quick-drench facilities in work area.

9. Physical and Chemical Properties

Appearance:
Colorless crystals.

Odor:
Odorless.

Solubility:
97g/100g water @ 25C (77F).

Specific Gravity:
2.43

pH:
Slightly acidic to litmus

% Volatiles by volume @ 21C (70F):
0

Boiling Point:
Not applicable.

Melting Point:
452C (846F)

Vapor Density (Air=1):
No information found.
Vapor Pressure (mm Hg):
1 @ 198°C (388°F)
Evaporation Rate (BuAc=1):
No information found.

10. Stability and Reactivity

Stability:
Stable under ordinary conditions of use and storage. Slightly hygroscopic; slowly becomes yellowish in air.

Hazardous Decomposition Products:
Emits toxic fumes of nitric oxides, ammonia, hydrogen bromide, and bromine when heated to decomposition.

Hazardous Polymerization:
Will not occur.

Incompatibilities:
Bromine trifluoride, iodine heptafluoride, potassium, and acid salts. Reacts with acids to form toxic hydrogen bromide and bases to form ammonia.

Conditions to Avoid:
Moisture and incompatibles.

11. Toxicological Information

Oral rat LD50: 2700 mg/kg

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>NTP Carcinogen Known</th>
<th>Anticipated</th>
<th>IARC Category</th>
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<td>Ammonium Bromide / (NH4)Br/</td>
<td>No</td>
<td>No</td>
<td>None</td>
</tr>
</tbody>
</table>

12. Ecological Information

Environmental Fate:
No information found.

Environmental Toxicity:
No information found.

13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

14. Transport Information

Not regulated.
15. Regulatory Information

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Chemical Inventory Status - Part 1

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<th>Ingredient</th>
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Chemical Inventory Status - Part 2

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Federal, State & International Regulations - Part 1

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<th>RQ</th>
<th>TPQ</th>
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Federal, State & International Regulations - Part 2

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<td>No</td>
<td>No</td>
</tr>
</tbody>
</table>


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Australian Hazchem Code: None allocated.
Poison Schedule: None allocated.
WHMIS:
This MSDS has been prepared according to the hazard criteria of the Controlled Products Regulations (CPR) and the MSDS contains all of the information required by the CPR.

16. Other Information

NFPA Ratings: Health: 2  Flammability: 0  Reactivity: 0
Label Hazard Warning:
WARNING! HARMFUL IF SWALLOWED OR INHALED. AFFECTS CENTRAL NERVOUS SYSTEM, BRAIN AND EYES. MAY CAUSE IRRITATION TO SKIN, EYES, AND RESPIRATORY TRACT.
Label Precautions:
Avoid breathing dust.
Keep container closed.
Use with adequate ventilation.
Avoid contact with eyes, skin and clothing.
Wash thoroughly after handling.
Label First Aid:
In all cases call a physician. If swallowed, induce vomiting immediately as directed by medical personnel. Never give anything by mouth to an unconscious person. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. In case of contact, immediately flush eyes or skin with plenty...
of water for at least 15 minutes.

**Product Use:**
Laboratory Reagent.

**Revision Information:**
No Changes.

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